Breakfast

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| **Across****2.** abbreviation for atomic mass unit**5.** a change that occurs without altering the chemical composition**6.** is a comparison of how much matter there is in a certain amount of space. It is a ratio of mass and volume.**10.** a change that results in the formation of a new chemical substance through the making or breaking of bonds between atoms.**11.** a measurement of the amount of matter something contains. It is typically measured in grams.**13.** is the amount of space that an object or substance occupies. It is typically measured in cm^3 or mL.**14.** negative charge, subatomic particle found in the nucleus**16.** formed by physically mixing substances together. Retain their own properties and can be separated out by physical means.**17.** particles such as electrons, protons, and neutrons**19.** small particles that make up matter**20.** things that make up the periodic table | **Down****1.** protons + neutrons =**3.** composed of two or more atoms of the same element. These atoms are chemically bonded together. **4.** neutral charge, subatomic particle found in the electron cloud**7.** found outside of nucleus, made up of space and electrons**8.** substance made up of two or more elements chemically combined**9.** a substance that cannot be broken down any further by chemical or physical means**12.** found in the center of the atom**15.** a substance or material which occupies space and possess rest mass**18.** positive charge, subatomic particle found in the nucleus |