Breakfast

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| **Across**  **2.** abbreviation for atomic mass unit  **5.** a change that occurs without altering the chemical composition  **6.** is a comparison of how much matter there is in a certain amount of space. It is a ratio of mass and volume.  **10.** a change that results in the formation of a new chemical substance through the making or breaking of bonds between atoms.  **11.** a measurement of the amount of matter something contains. It is typically measured in grams.  **13.** is the amount of space that an object or substance occupies. It is typically measured in cm^3 or mL.  **14.** negative charge, subatomic particle found in the nucleus  **16.** formed by physically mixing substances together. Retain their own properties and can be separated out by physical means.  **17.** particles such as electrons, protons, and neutrons  **19.** small particles that make up matter  **20.** things that make up the periodic table | **Down**  **1.** protons + neutrons =  **3.** composed of two or more atoms of the same element. These atoms are chemically bonded together.  **4.** neutral charge, subatomic particle found in the electron cloud  **7.** found outside of nucleus, made up of space and electrons  **8.** substance made up of two or more elements chemically combined  **9.** a substance that cannot be broken down any further by chemical or physical means  **12.** found in the center of the atom  **15.** a substance or material which occupies space and possess rest mass  **18.** positive charge, subatomic particle found in the nucleus |