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Intro to Chemistry

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| **Across****4.** Rows on the Periodic Table**7.** A type of chemistry that involves carbon based life forms.**9.** positively charged particle.**12.** Chemistry that relates to non carbon compounds.**13.** How quick it is to join up with other elements.**14.** Element with 2 protons.**15.** A negatively charged particle. **17.** The place where Protons and Neutrons are housed**18.** A particle that has no charge.**19.** Number of Protons in an Element**22.** Same number of protons different number of neutrons. **23.** Diagram of Elements**24.** Creator of the Periodic Table | **Down****1.** When two elements bond together. **2.** The lightest element known**3.** Point at which a liquid is turned into a gas.**5.** A substance with just one type of atom.**6.** A type of mixture that you can pick apart.**8.** A type of mixture that you cannot pick apart. A solution.**10.** The point at which a solid turns to a liquid.**11.** Al things that have mass and volume.**16.** Electrons are located around the nucleus in an electron what? **20.** When two or more compounds are joined together.**21.** Group 18. These elements do not mix well with others.  |