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Solutions Vocabulary

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| **Across****1.** The quality or property of being soluble**4.** holding as much water or moisture as can be absorbed; thoroughly soaked**7.** the act of concentrating; the state of being concentrated.**11.** A substance composed of two or more metals, or of a metal or metals with a nonmetal**13.** to make (a liquid) thinner or weaker by the addition of water**14.** The minor component in a solution, dissolved in the solvent.**15.** the stage at which no more of a substance can be absorbed into a vapor or dissolved into a solution**16.** The liquid in which a solute is dissolved to form a solution.**17.** A liquid mixture in which the minor component (the solute) is uniformly distributed within the major component (the solvent).**18.** any combination of substances that has uniform composition and properties**19.** A sample of matter with both definite and constant composition with chemical properties | **Down****2.** a mixture of physically distinct substances with different properties**3.** A substance made by mixing other substances together.**5.** Incapable of being dissolved or liquefied**6.** denoting fats containing a high proportion of fatty acid molecules with at least one double bond**8.** An example is milk or mayo**9.** Is composed of two or more separate elements; a mixture.**10.** A substance consisting of atoms which all have the same number of protons and the same atomic number.**12.** Capable of being dissolved or liquefied |