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Solutions Vocabulary

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| **Across**  **1.** The quality or property of being soluble  **4.** holding as much water or moisture as can be absorbed; thoroughly soaked  **7.** the act of concentrating; the state of being concentrated.  **11.** A substance composed of two or more metals, or of a metal or metals with a nonmetal  **13.** to make (a liquid) thinner or weaker by the addition of water  **14.** The minor component in a solution, dissolved in the solvent.  **15.** the stage at which no more of a substance can be absorbed into a vapor or dissolved into a solution  **16.** The liquid in which a solute is dissolved to form a solution.  **17.** A liquid mixture in which the minor component (the solute) is uniformly distributed within the major component (the solvent).  **18.** any combination of substances that has uniform composition and properties  **19.** A sample of matter with both definite and constant composition with chemical properties | **Down**  **2.** a mixture of physically distinct substances with different properties  **3.** A substance made by mixing other substances together.  **5.** Incapable of being dissolved or liquefied  **6.** denoting fats containing a high proportion of fatty acid molecules with at least one double bond  **8.** An example is milk or mayo  **9.** Is composed of two or more separate elements; a mixture.  **10.** A substance consisting of atoms which all have the same number of protons and the same atomic number.  **12.** Capable of being dissolved or liquefied |